## 9-BORABICYCLO[3.3.1]NONANE INDUCED FRAGMENTATION OF 2-AZANORBORNENES: A FORMAL BORA-AZA RETRO ENE REACTION

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Abstract: N-Substituted 2-azanorbornenes undergo a 9-borabicyclo[3.3.1]nonane induced fragmentation in tetrahydrofuran giving rise to cyclopentenylmethyl amines (cf 3 and 5).

It has been demonstrated that 2-azanorbornenes undergo facile acid-catalyzed retro Diels-Alder reaction under a variety of conditions giving rise to primary amines<sup>2</sup> or N-methylated amines.<sup>3</sup> We wish to report that the retro Diels-Alder process can be interrupted leading to the formation of cyclopentenylmethyl amines of the type illustrated in Equation 1 by complexation of 2-azanorbornenes with 9-borabicyclo[3.3.1]nonane(9-BBN).



The 9-BBN mediated fragmentation of azanorbornenes, which constitutes a formal bora-aza retro ene reaction, was first observed during an attempt to hydroborate N-benzyl-2-azanorbornene 1 ( $R = CH_2C_6H_5$ ). Exposure of 1 ( $R = CH_2C_6H_5$ ) to an equivalent of 9-BBN at amblent temperature provided less than 10% of the anticipated hydroboration products. The major product (44%) was cyclopentenylmethyl amine 3 ( $R = CH_2C_6H_5$ ). Approximately 25% of 1 ( $R = CH_2C_6H_5$ ) was recovered. No reaction was observed at 0°C. The formation of 3 is best performed in tetrahydrofuran at reflux. For example, exposure of a 0.1 M solution of N-benzyl-2-azanorbornene in tetrahydrofuran to 1.0 equiv of 9-BBN at reflux over 17 h provides the corresponding cyclopentenylmethyl amine 3 ( $R = CH_2C_6H_5$ ) in 54% yield along with 20% of recovered

starting material. Use of 1.2 equiv of 9-BBN did not improve the yield. After the fact, the formation of 3 (R =  $CH_2C_6H_5$ ) is not unreasonable in view of the well-documented<sup>4</sup> affinity of boron for amines and our observation that protonation<sup>2,3</sup> of the nitrogen atom in 2-azanorbornenes dramatically weakens the C(1), N(2) carbon nitrogen bond.

The boron mediated ring opening detailed above has been applied to a number of N-substituted 2azanorbornenes (Table I). All reactions were performed in tetrahydrofuran at elevated temperatures employing 1.0 equiv of 9-BBN. Yields are modest in the range of 50-60%. Efforts to improve yields by employing diborane or disiamylborane failed to give any ring cleaved product. Use of diborane provided only hydroboration products. Attempts to extend this chemistry to N-benzyl-2-azabicyclo[2.2.2]octene resulted in complete recovery of starting material with no trace of ring opened product.



The moderate success achieved above with unsubstituted 1-azanorbornenes led us to investigate more functionalized azanorbornenes. In this regard a number of 3-substituted 2-azanorbornenes were prepared<sup>5</sup> and examined. In a preliminary study, azanorbornene 4 was exposed to 1.0 equiv of 9-BBN in tetrahydrofuran at 50°C. After 22 h, only a 12% yield of 5 was obtained. The yield of 5 could be improved by



employing 2.0 equiv of 9-BBN in tetrahydrofuran at 50°C. After 1 h, the starting material was completely consumed giving rise to a 44% yield of 5. The above conditions were applied to a number of 3-substituted azanorbornenes (Table II). Inspection of the data reveals that the exo-substituted azanorbornenes provided



modest yields (44-59%) of the corresponding cyclopentenylamines whereas the endo-substituted azanorbornenes gave rise to poor yields of rearranged products. The low yields associated with the 3-endo substituted substrates may be due to severe steric crowding on the endo face of the azanorbornene which inhibits complexation. As a consequence one observes hydroboration of the olefin.

The rearrangement of 2-azanorbornenes with 9-BBN reported above can be derived from zwitterionic intermediate 2 via two different pathways: (a) a concerted shift of hydride from boron to C(5) or (b) initial bond cleavage of the C(1), N(2) bond leading to allyl cation 6 which subsequently is trapped by transfer of hydride from boron.



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## References

- 1. Present address: ICOS Corporation, 22021 20th Avenue SE, Bothell, WA 98021
- Grieco, P.A.; Parker, D.T.; Fobare, W.F.; Ruckle, R. J. Am. Chem. Soc., 1987, 109, 5859. Grieco,
  P.A.; Clark, J.D. J. Org. Chem., 1990, 55, 2271.
- 3. Grieco, P.A.; Bahsas, A. J. Org. Chem., 1987, 52, 5746.
- 4. Greenwood, N.N.; Earnshaw, A. Chemistry of the Elements, Pergamon Press, New York, 1984.
- 5. Grieco, P.A.; Larsen, S.D.; Fobare, W.F. Tetrahedron Lett., 1986, 27, 1975.

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